

Inhibitory Activity of Silver Nanoparticles and Sodium Hypochlorite against Biofilm Produced by Salmonellae Isolated from Poultry Farms

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ARTICLE INFO

Original Research

Received:

07 July 2019

Accepted:

22 September 2019

Keywords:

Biofilm; Poultry farms;
Salmonella;
Silver nanoparticles;
Sodium hypochlorite

ABSTRACT

Salmonella Typhimurium and *Salmonella* Enteritidis are among the predominant *Salmonella* serotypes in the Egyptian poultry farms. *Salmonella* has the ability to build up biofilms on a variety of surfaces. The antibiofilm activities of silver nanoparticles (AgNPs) and sodium hypochlorite (NaOCl) on prevention and controlling of biofilm by *Salmonella* spp. was estimated. Silver nanoparticles exhibited bactericidal activity against both *S. Typhimurium* and *S. Enteritidis* with MIC value at 15 µg/ml, while, that of NaOCl was 1600 µg/ml. AgNPs (25 µg/ml) could inhibit biofilm formation at percentages of 84.96% and 78.85% against *S. Typhimurium* and *S. Enteritidis*, respectively. A percentage of 87 % biofilm removal by AgNPs after 3 h contact with the built-up biofilm produced by *S. Typhimurium* and *S. Enteritidis* was recorded. NaOCl (2200 µg/ml) exhibited inhibition percentages of biofilm formation at 83.89% and 75.76% against *S. Typhimurium* and *S. Enteritidis*, respectively. While, biofilm removal percentages after 2 h contact between NaOCl (2200 µg/ml) with the formed biofilm by *S. Typhimurium* and *S. Enteritidis* were 87.42% and 89.37%, respectively. It can be concluded that AgNPs and NaOCl were able to promote a significant reduction of biofilm formation by *S. Typhimurium* and *S. Enteritidis*. Also, AgNPs and NaOCl effectively oppress the mature biofilms formed and the antibiofilm efficiency increased with the increase of contact time with the biofilms.

J. Adv. Vet. Res. (2019), 9 (4), 151-160

Introduction

Bacterial biofilm is an association of micro-organisms within a self-produced matrix of extracellular polymeric substance, in which bacterial cells adhere to each other on living or non-living surfaces. It is infectious in nature and can result in various infections. Moreover, biofilm formation protects bacterial cells against a wide range of challenges including UV light radiation, pH and osmotic changes, dehydration, host immune responses, antimicrobial agents and disinfectants (Nilsson *et al.*, 2011; Bogino *et al.*, 2013). The established biofilms can tolerate antimicrobial agents at concentrations of 10-1000 times that needed to kill planktonic bacteria and also have an extraordinary resistance to phagocytosis, which make it difficult to eradicate biofilms from living hosts (Cos *et al.*, 2010). Therefore, a greater understanding of bacterial biofilm is required for the development of novel, effective control strategies that can overcome the drawbacks of the current control ones. One of these novel alternatives could be metal-based nanoparticles that have been utilized in several applications due to their fast and broad antibacterial activity as well

as low production costs (Fabrega *et al.*, 2011).

The antimicrobial activity of silver nanoparticles (AgNPs) in reducing the growth of different microorganisms is well documented (Li *et al.*, 2006; Kanematsu *et al.*, 2009). AgNPs are now considered one of the most promising strategies to control bacterial infections. The bactericidal efficiency of AgNPs has been attributed to their small size and high surface to volume ratio, which allows them to interact closely with microbial membranes (Morones *et al.*, 2005; Rai *et al.*, 2009). Furthermore, AgNPs have not been shown to cause bacterial resistance because silver nanoparticles exert their antibacterial effects at several sites (Shrivastava *et al.*, 2007). Smaller AgNPs can reduce more biomass and viability of biofilms, due to better penetration into the exopolysaccharide (EPS) matrix (Habash *et al.*, 2014). Moreover, the anti-biofilm activity of AgNPs could be due to inhibition of EPS synthesis in bacteria, which limits the biofilm formation (McLaughlin-Borlace *et al.*, 1998).

Sodium hypochlorite (NaOCl) is one of the most effective disinfectants against biofilms (Pui *et al.*, 2011; Rodrigues *et al.*, 2011); with the ability to eradicate biofilms at concentrations as low as 3.125 mg per ml (Rodrigues *et al.*, 2011). During treatment, sodium hypochlorite decomposes to sodium hydroxide and hypochlorite, which is a strong oxidizing agent (Tote *et al.*, 2010) that act by oxidizing the bacterial cell mem-

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brane resulting in cell lysis and death (Maris, 1995; Hawkins and Davies, 1999).

In Egypt, poultry production represents one of the most important economic and animal protein sources for human, however this sector faces many problems and challenges. One of these problems is the infection with various microbes such as *Salmonella*. *Salmonella enterica* serovar Typhimurium and *Salmonella enterica* serovar Enteritidis have been identified as the predominant serotypes present in Egyptian poultry farms (Abd El-Ghany et al., 2012). There are many reports regarding the development of biofilms by *Salmonella* on a variety of biotic and abiotic surfaces, where they allow *Salmonella* to survive and spread in the environment (Janssens et al., 2008). Furthermore, *Salmonella* contamination on farms may come from a variety of sources, including wild animals, rodents, insects, feed and human (Meerburg and Kijlstra, 2007; Carrique-Mas et al., 2009; Hilbert et al., 2012). Control of *Salmonella* is a challenge to public health because of their emergence/ re-emergence and high mutation rate, with antibiotic resistance in both developed and developing countries (Jassim and Limoges, 2017). *Salmonella* found in biofilms showed a higher tolerance to antibiotics and was less susceptible to disinfectants than planktonic *Salmonella* (Pui et al., 2011; Sheffield and Crippen, 2012). The use of glutaraldehyde, formaldehyde and peroxygen at a concentration of 1% in field conditions is insufficient to eradicate *Salmonella* biofilms (Marin et al., 2009).

High prevalence of *Salmonella* spp. in water troughs swabs and water was reported by Ogden et al. (2007) and Marin et al. (2009), where many strains of *Salmonellae* were able to produce biofilm in water tanks, drinker lines and stainless-steel surfaces, and continuously release planktonic cells in water. Therefore, this biofilm is one of the potential and continuous sources of infection because it helps salmonellosis to be transferred to the entire flock. So, to achieve efficient control of *Salmonella* infection in poultry farms there is a need for an effective method for removal and prevention of biofilm formation on these surfaces.

The misuse of antibiotics and different antimicrobials in the Egyptian poultry industry led to the emergence of resistant strains that are difficult to eradicate. Therefore, we need to evaluate the activity of other alternative agents against the strains isolated from Egyptian poultry farms to be able to achieve good hygienic measures and control biofilm producing bacteria in both poultry farms as well as human establishment. The objective of the current study was to evaluate the effectiveness of AgNPs and NaOCL on prevention and controlling of biofilm formation by *Salmonella* spp. isolated from Egyptian poultry farms.

Materials and methods

Bacterial strains

The bacterial strains used in the current study were iso-

lated from 11 poultry farms located in Assiut Governorate, Egypt. A total of 600 samples were collected from the poultry farms including litter, feed, water samples, as well as water troughs, cloacal and wall swabs. *Salmonella* detection was carried out according to standard methods (Anon, 2007), the samples were initially transferred into buffered peptone as a pre-enrichment medium and incubated for 18-20 h at 37°C then 100 µl of the pre-enriched cultures was transferred to Rappaport-Vassiliadis broth (Biolife EM1804) and incubated at 42°C. After 24 h of incubation, one loopful of enriched broths was streaked onto plates of xylose lysine deoxycholate agar (Biolife EG8702) and incubated at 37°C for 24-48 h. The plates were examined for the presence of typical colonies of *Salmonella* spp. i.e. red colonies with black centers (Antunes et al., 2003). Pure colonies were subjected to identification using gram staining, biochemical reactions and finally multiplex PCR for detection of *S. Typhimurium* and *S. Enteritidis* that was performed in Molecular Biology Unit, Assiut University, Egypt. *S. Typhimurium* strain was obtained from Animal Health Research Institute, Giza, Egypt and served as a positive control during multiplex PCR.

As shown in Table 1, the primers' sets used were ST11 and ST15 as a universal gene for *Salmonella* spp., S1 and S4 for *S. Enteritidis*, and Fli15 and Typ04 for *S. Typhimurium* according to Soumet et al. (1999). Extraction was carried out using Patho Gene DNA/RNA extraction kit (iNtRON Biotechnology) according to the manufacture' instructions. Multiplex PCR reaction was done in a total volume of 22 µl containing 11 µl PCR master mix, 1 µl of each primer and 5 µl of the extracted DNA. The thermocycler conditions included initial denaturation at 95 °C for 5 min., followed by 35 cycles of 95 °C for 1 min., 48 °C for 1 min., and 72 °C for 1 min., with final extension at 72 °C for 10 min., after amplification, 1% agarose was used for electrophoresis.

The ability of each bacterial strain to produce biofilm in pure culture was determined using the tissue culture plate method according to Coffey and Anderson (2014). One to two colonies from fresh overnight cultured agar plates of each strain were transferred to 5 ml tryptic soy broth (TSB) and the bacteria was allowed to grow at 37 °C for 18 h. Then a 1:100 dilution was prepared in TSB using 0% glucose and 2% glucose as a supplement (Ganjali Dashti et al., 2016) and following thorough mixing, 100 ul was transferred to each well in 96 microtiter tissue culture plates then incubated at 28°C without shaking for 48 h. After that the liquid media from each well was gently removed and the wells were washed three times with distilled water to remove the free-living bacterial cells that do not contribute to biofilm formation. The wells were stained with 125 µl of 0.1% crystal violet for 15-30 min. to confirm the presence of biofilm. Excess stain was rinsed off by thorough washing with distilled water three times, and then the plates were kept for 30 min till dryness. The microtiter plate was examined for the presence of purple ring that indicating the biofilm formation. For biofilm quantification 150 µl 30% acetic acids were added to each well and kept for 10 min-

Table 1. Oligonucleotide sequences used to detect *Salmonella* spp. (Soumet et al., 1999)

Primer sets	Type	Primer sequence 5'	3'	Amplification product (bp)
<i>Salmonella</i> spp.	ST11(1)	GCC AAC CAT TGC TAA ATT GGC GCA		429
	ST15(1)	GGT AGA AAT TCC CAG CGG GTA CTG G		
<i>S. Enteritidis</i>	S1 (2)	GCC GTA CAC GAG CTT ATA GA		250
	S4 (2)	ACC TAC AGG GGC ACA ATA AC		
<i>S. Typhimurium</i>	Fli15(3)	CGG TGT TGC CCA GGT TGG TAA T		620
	Typ04 (3)	ACT GGT AAA GAT GGC T		

utes and the contents of each well were mixed by repeated pipetting, and then 125 µl of the solubilized solution was transferred to a new well of a new optically clear flat bottom 96-well plate. The optical density (OD) of each well was measured at a wavelength 545 nm using Microtiter plate reader (Model Start Fax-2100, Awareness Technology INC).

Efficiency of AgNPs and NaOCL against S. Enteritidis and S. Typhimurium biofilm Preparation of silver nanoparticles (AgNPs)

Stable AgNPs less than 100 nm was synthesized in a typical one-step protocol according to Vigneshwaran et al. (2006); where, one gram of soluble starch was added to 100 ml of deionized water and heated till complete dissolution, then 1 ml of a 100 mM aqueous solution of silver nitrate (AgNO₃), was added and stirred well. This mixture was put into dark glass bottle and kept in an autoclave for 5 min at 121°C. The resulting solution was clear yellow in color indicating the formation of AgNPs. After preparation of silver nanoparticles, the stock solution was kept away from direct sunlight at room temperature. The concentration and size of the particles were measured before its use. The total concentration of AgNPs was measured by Graphite Furnace Atomic Absorption (Model 210VGP) at the Faculty of Science, Assiut University, Egypt. The size of the particles was measured by Transmission Electron Microscope (TEM) (JEOL-JEM- 100CX II) at Electron Microscopy Unit, Assiut University, Egypt.

Determination of minimum inhibitory concentration (MIC) and minimum bactericidal concentration (MBC) of AgNPs and NaOCL

The MIC of AgNPs and NaOCL against *S. Typhimurium* and *S. Enteritidis* was measured by standard broth macro-dilution method as described by Sharma et al. (2015). Briefly, 1-2 overnight fresh colonies were added to 7 ml TSB and incubated for 18 h at 37°C. Different concentrations of AgNPs (5, 10, 15, 20, 25, 30 µg/ml) and NaOCL (5% available chlorine) (1000, 1200, 1400, 1600, 1800, 2000, 2200, 2400, 2600 and 2800 µg/ml) were separately added to each tube containing 5 ml TSB and finally an aliquot of bacterial culture (5 × 10⁵ CFU ml⁻¹) were added separately to these tubes. After that, the tubes were incubated at 37°C for 18 h. In the control positive set, only the organisms were grown in absence of AgNPs and NaOCL, while tubes that contain TSB plus AgNPs and NaOCL in absence of inoculum were considered the control negative set. The MIC was considered as the lowest concentration of AgNPs and NaOCL in which there was no visible bacterial growth after 18 h of incubation at 37°C. To determine the MBC, the bacterial broth at the lowest concentration, which give no visible turbidity as well as at two higher concentrations were sub-cultured to sterile petri dishes containing standard plates count agar in duplicated to assure the absence of colony forming units (CFU). The petri dishes without any bacterial colonies were considered positive for the MBC.

Prevention and eradication of Salmonella biofilm

The ability of AgNPs and NaOCL to prevent biofilm formation and eradicate the established biofilm produced by *Salmonellae* was estimated according to Abidi et al. (2014). A total number of 12 strains including *S. Typhimurium* (n.=10) and *S. Enteritidis* (n.=2) were used in this experiment. Glucose supplement (2%) was applied during bacterial growth for biofilm production. At first the anti-biofilm activities of AgNPs and NaOCL were evaluated during bacterial incubation; while the biofilm was being created. Briefly, 1-2 colonies from fresh overnight cultured agar plates of each strain were transferred to 5 ml TSB and allowed to grow at 37°C for 18 h. Then the

culture was diluted 1:100 in TSB with glucose supplement and 100 µl of diluted culture was pipetted in microtiter plate wells. After that, 100 µl of each concentration of AgNPs or NaOCL was added into three wells and then plate was covered and incubated at 28 °C for 48 h. Each bacterial strain was treated with four concentrations of AgNPs (10, 15, 20, and 25µg/ml) and four concentrations of NaOCL (1600, 1800, 2000 and 2200 µg/ml). Furthermore, for each strain, three wells of the 96-well flat bottom microtiter plate were inoculated with bacterial inoculums without treatments (positive control) and another three wells for negative control (treatment with TSB only). Concerning the second experiment, anti-biofilm activity of AgNPs and NaOCL was estimated after incubation; on the established biofilm. Briefly, microtiter plates were inoculated as mentioned above and incubated for biofilm production. After incubation, plates were washed with sterile water to remove planktonic cells after that 200 µl of each tested agent (25 µg/ml AgNPs and 2200 µg/ml NaOCL) was transferred into each well, with exception of blank and positive control wells. The plates were incubated for different contact times, namely (30, 60, 90, 120, 150 and 180 min.) for AgNPs and (15, 30, 45, 60, 90 and 120 min.) for NaOCL. At the end of each contact time, AgNPs and NaOCL were quenched by adding 5 g/l sodium thiosulfate (Na₂S₂O₃) to stop their antimicrobial reaction as described in the European Quality Standards (NEN-EN 1276, 1997). Following the desired incubation time, the planktonic bacterial cells were removed from the microtiter plates and the wells were washed several times. Subsequently, for biofilm staining, 125 ul of 0.1% crystal violet solution was added to each well and incubated for 10 min. at room temperature. Following incubation, the stain was removed, and plates were washed with vigorous shaking to remove all liquid. Subsequently, the plates were inverted and vigorously tapped on paper towels to remove all the contents and left to air dry. Finally, the dye was solubilized by adding 150 µl of 30% acetic acid to each well of the plate, and the plate was incubated for 10-15 minutes at room temperature. In the next step, contents of each well were mixed by repeated pipetting and then 125 µl of the crystal violet-acetic acid solution was transferred from each well to a separate well of a new optically clear flat-bottom 96-well plate. Optical density (OD) of each of these 125-µl samples was measured at a wavelength 545 nm., and to measure the antibiofilm efficacy; the reduction/ removal percentages were calculated using the following equation according to Abidi et al. (2014):

$$\text{Reduction/Removal Percentage} = [(C-B) - (T-B) / (C-B)] * 100\%$$

Where B = absorbance of blank (no biofilm, no treatment).

C = absorbance of control (biofilm, no treatment).

T = absorbance of test (biofilm and treatment).

Statistical Analysis

Statistical analysis of data was carried by using SPSS software version 17. The data was subjected to analyses of variance using the ANOVA procedure and General Linear Models Procedure (GLM procedure) of SPSS software. The results of optical densities were presented as mean and standard deviations (SD) for each variable. Significant differences between mean values were tested using Duncan's multiple range test. P-value was considered statistically significant when P < 0.05.

Results

Isolation of Salmonella from poultry farms and the ability to build up biofilm

Salmonellae were isolated from 11 broiler and layer farms

in Assiut Governorate, Egypt. Thirty-Five biochemically positive *Salmonella* strains were subjected to molecular identification by multiplex PCR, from which 19 strains were positive for *S. Typhimurium* and two strains were positive for *S. Enteritidis*. Data illustrated in Fig. 1, represents the molecular identification of *S. Typhimurium* and *S. Enteritidis* by multiplex PCR. The data presented in Table 2, revealed that 54.29% of the strains were positive for *S. Typhimurium* with the highest percentage (26.32%) was recorded for litter samples followed by feed and water troughs swabs. Furthermore, 5.71% of the examined strains were positive for *S. Enteritidis*, representing 50 % for each cloacal and wall' swab.

The ability of 21 *Salmonella* strains (*S. Typhimurium* (n.=19) and *S. Enteritidis* (n.=2) strains) for biofilm formation was checked and the result revealed 10 strains *S. Typhimurium* and the two strains *S. Enteritidis* were able to produce biofilm.

MIC and MBC of AgNPs or NaOCl on S. Typhimurium and S. Enteritidis

As shown in Fig. 2, the size of AgNPs used in the current study was ranged from 11.7 to 18.9 nm with spherical shape. AgNPs exhibited bactericidal activity against *S. Typhimurium* and *S. Enteritidis* where the MIC value was 15 µg/ml. While, the lowest concentration that prevent the bacterial growth on plate count agar (MBC) was 20 µg/ml. MIC and MBC of NaOCl on both *S. Typhimurium* and *S. Enteritidis* was 1600 µg/ml and 2000 µg/ml, respectively.

Prevention of S. Typhimurium and S. Enteritidis biofilm formation in the presence of AgNPs or NaOCL

As shown in Fig. 3A and B, the mean values of optical densities of bacterial cultures from quantified biofilm formed by

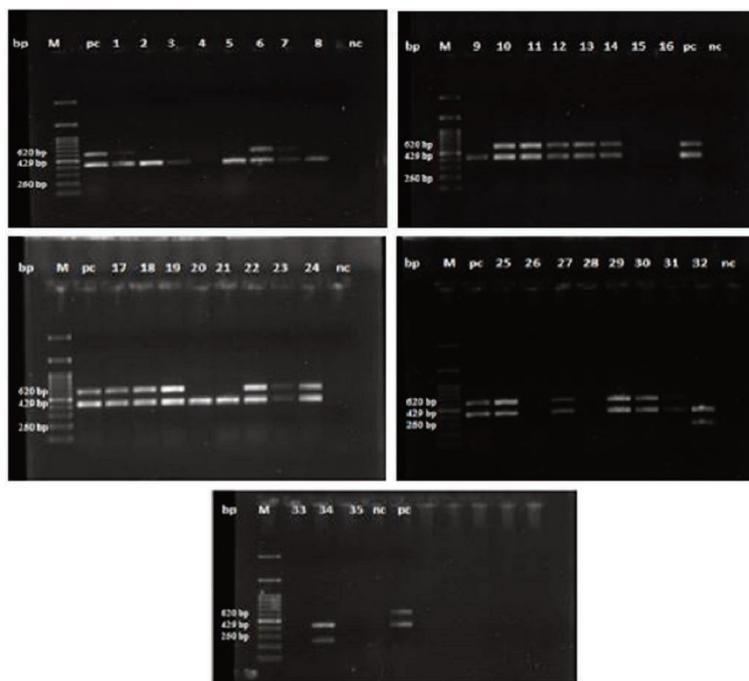


Fig. 1. Multiplex PCR identifying *S. Typhimurium* and *S. Enteritidis*. M:100 bp DNA marker; pc: positive control; nc: negative control; Lanes 2,3, 5, 8, 9, 20 and 21 are positive amplification of 429 bp fragments of *Salmonella* spp., Lanes 1, 6, 7, 10, 11, 12, 13, 14, 17, 18, 19, 22, 23, 24, 25, 27, 29, 30 and 31 are positive amplification of 620 bp fragments of *Salmonella* Typhimurium. Lanes 32 and 34 are positive amplification of 250 bp fragments of *Salmonella* Enteritidis.

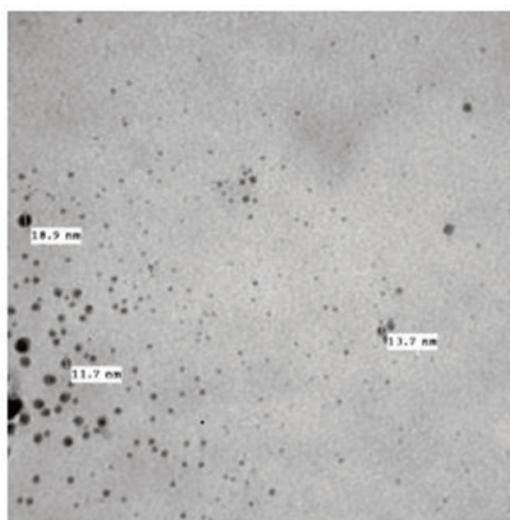


Fig. 2. Transmission Electron Microscope of silver nanoparticles (AgNPs) showing AgNPS spherical in shape with size ranged from 11.7 to 18.9 nm.

Table 2. Molecular identification of *S. Typhimurium* and *S. Enteritidis* by multiplex PCR

	Sample type												Total no. of positive PCR samples (n.=21)	
	Litter samples (n.=10)		Feed samples (n.=6)		Water samples (n.=5)		Water troughs swabs (n.=5)		Cloacal swabs (n.=5)		Wall swabs (n.=4)			
	No.	%	No.	%	No.	%	No.	%	No.	%	No.	%		
<i>Salmonella Ty-</i> <i>phimurium</i>	5	26.32	4	21.05	2	10.53	4	21.05	3	15.79	1	5.26	19	54.29
<i>Salmonella Enteritidis</i>	0	0	0	0	0	0	0	0	1	50	1	50	2	5.71

S. Typhimurium and *S. Enteritidis* were significantly reduced in all AgNPs exposed bacterial cells at concentration of 10, 15, 20, 25 µg/ml when compared with the control ($P < 0.05$). Furthermore, there were no significant differences between the mean values of optical densities of bacterial culture exposed to the various concentrations of silver nanoparticles. Moreover, the highest inhibition percentages; 84.96% and 78.85% of biofilm formation by *S. Typhimurium* and *S. Enteritidis* were observed when the bacterial cells treated with 25 µg/ml of AgNPs, respectively.

The obtained data revealed that the mean values of optical densities of bacterial culture from the formed biofilm by *S. Typhimurium* and *S. Enteritidis* were significantly reduced in all NaOCL exposed bacterial culture when compared with the control ($P < 0.05$) as presented in Fig. 4A, B. Furthermore, there were no significant differences among the mean values of optical densities of bacterial culture subjected to various concentration of NaOCL. When calculating the inhibition percentages of biofilm formation by *S. Typhimurium* and *S. Enteritidis* treated with 1600 µg/ml, 1800 µg/ml, 2000 µg/ml, 2200 µg/ml NaOCL, the highest percentages; 83.89% and 75.76% were observed at the highest used concentration, respectively.

Control of biofilm formed by *S. Typhimurium* and *S. Enteritidis* by AgNPs (25 µg/ml) or NaOCL (2200 µg/ml)

S. Typhimurium and *S. Enteritidis* established biofilms were exposed to AgNPs (25 µg/ml) for different contact times and the obtained data revealed that there were significant differences between the effect of AgNPs at all contact times and the control ($P < 0.05$) as presented in Fig. 5. Moreover, there were significant differences between the mean values of optical densities for *S. Typhimurium* at 2.5 h, 3 h and the mean values at 30 min., 1 h, 1.5 h, and there were significant differences between the mean values at 2 h and the means at 30 min, and 1 h ($P < 0.05$). The percentage of biofilm removal by 25 µg/ml AgNPs were 50.85%, 60.20%, 71%, 81.41%, 84.60% and 87.43% after contact time of 30 min, 1 h, 1.5 h, 2 h, 2.5 h, 3 h, respectively. For *S. Enteritidis* the results revealed that the biofilm removal percentage after the examined contact times were 48.70%, 63.15%, 78.93%, 80.94%, 83.63% and 87.67 %, respectively. Additionally, there were significant differences between all contact times and the control ($P < 0.05$). Meanwhile, there were significant differences between the mean values at 2.5 h, 3 h, and the mean at 30 min, 1 h, 1.5 h, 2 h, and there were significant differences between the mean at 1.5 h, 2 h, and that at 30 min ($P < 0.05$).

The percentages of biofilm removal by 2200 µg/ml NaOCL after various contact times for *S. Typhimurium* and *S. Enteritidis* were estimated with the highest percentage 87.42% and 89.37% after 2 h contact, respectively. The statistical analysis of data for both *S. Typhimurium* and *S. Enteritidis* showed that there were significant differences between all contact times and the control ($P < 0.05$), however, there were no significant differences among the different contact times as represented in Fig. 6 A, B.

Discussion

Salmonella remains an important concern in food processing environments as it causes salmonellosis, a major public health problem throughout the world (Anonymous, 2004). *S. Typhimurium* and *S. Enteritidis* were the most frequently identified *Salmonella* serovars in the Egyptian poultry farms and their surrounding environment (Abd El-Ghany et al. 2012; Oliveira et al., 2012), moreover, they were the predominant serotypes contributing biofilm production (Marin et al., 2009; Cevallos et al., 2012), for these reasons and from the hygienic

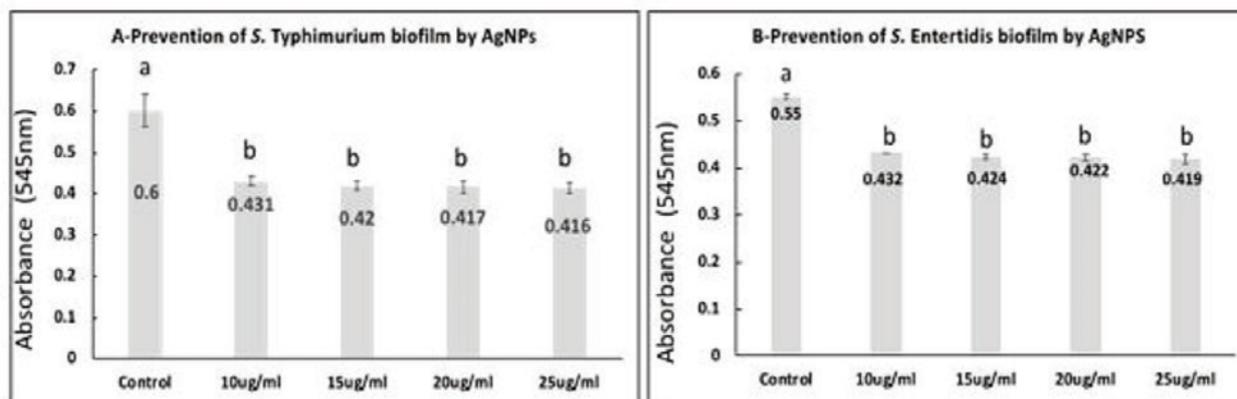


Fig. 3. *S. Typhimurium* (A) and *S. Enteritidis* (B) biofilms after 48 h incubation with the exposure to different concentrations of AgNPs. a,b: means with different letters are significantly different.

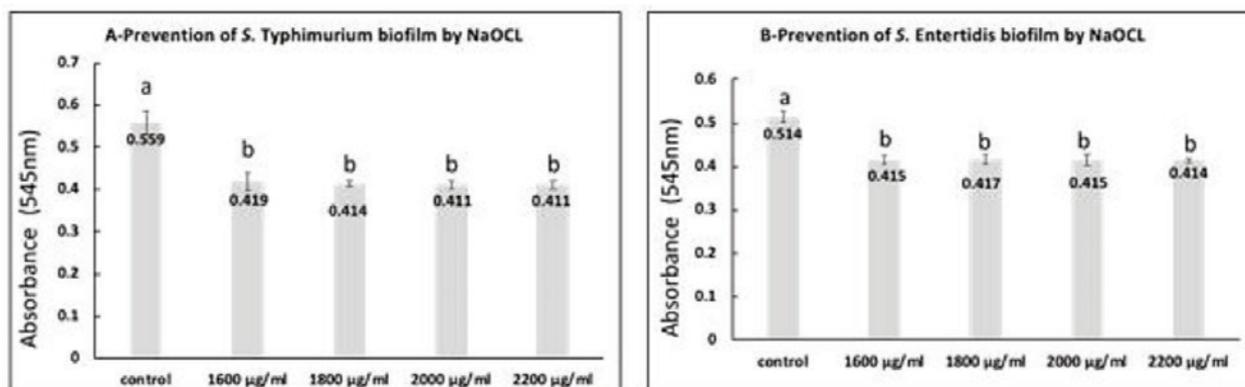


Fig. 4. *S. Typhimurium* (A) and *S. Enteritidis* (B) biofilms after 48 h incubation with the presence of different concentrations of NaOCl. a,b: means with different letters are significantly different.

point of view, these serovars were selected to study the antibiofilm activity of silver nanoparticles and sodium hypochlorite.

The obtained results revealed high prevalence of *Salmonella* spp. in litter samples, followed by water troughs swabs, cloacal swabs, feed samples, water samples and finally wall swabs. This finding agreed with Rodriguez et al. (2006), who reported that the prevalence of *Salmonella* in farm environments ranges from 10 to 26%. This can be explained by that faeces, soil, crevices, dusts, manure, litter, feeders and drinkers in farms and their environment harbor *Salmonella*, which increase the rate of contamination (Mallinson et al., 2000; Wales et al., 2006). Moreover, high prevalence of *Salmonella* spp. in litter is due to shedding of *Salmonella* from the intestinal tract of birds, which contaminates litter, bedding and feedstuff (Rodriguez et al., 2006; Hoelzer et al., 2011). The high prevalence of *Salmonella* spp. in water troughs swabs and water agreed with Murray (2000), who reported that *Salmonella* survives in water up to 56 days. Which may be attributed to failure to clean contaminated water or the water equipment after the removal of infected flock. Additionally, many strains of *Salmonella* spp. were able to produce biofilm in water tanks and drinker lines (Wilks et al., 2006; Ogden et al., 2007; Marin et al., 2009). Therefore, water is one of the potential sources of poultry infection because it helps salmonellosis to be transferred to the entire flock of the farm.

The ability of 21 *Salmonella* strains (*S. Typhimurium* (n.=19) and *S. Enteritidis* (n.=2) strains) for biofilm formation was checked and the result revealed 10 strains *S. Typhimurium* and the two strains *S. Enteritidis* were able to produce biofilm. Addition of glucose during biofilm production is much more supportive in biofilm formation as glucose provided the basic carbon units and the nutrient broth supplied basic support for

cell survival in biofilm (Coffey and Anderson, 2014; Ganjali Dashti et al., 2016). *Salmonella* is a hardy pathogen able to survive in many different environments (Carrique-Mas et al., 2009; Hilbert et al., 2012). In natural environments, bacteria exist in two forms; planktonic free bacteria are important for proliferation and spread of bacteria, or attached populations are necessary to allow the bacterial population to persist. During the last decades, it has become clear that bacteria, including food borne pathogens such as *Salmonella enterica*, grow predominantly as biofilms (Shemesh et al., 2007). Biofilm is a problem to the poultry industry, it is usually found in water systems of poultry farm and stainless steel in poultry processing plants and act as sources of cross-contamination (Wilks et al., 2006). Salmonellosis in chicken has been traced from biofilm associated with water tanks and drinker lines (Marin et al., 2009). Moreover, the transmission of *S. Typhimurium* between hosts, their survival in the environment and persistence in the host are facilitated by the ability to form biofilms (Cevallos et al., 2012), furthermore, bacterial biofilms are more resistant to disinfectants (Wong et al., 2010). From all these, we need to change how we deal with micro-organisms on dry surfaces, biofilm (Yezli and Otter, 2012; Donskey, 2013).

MIC and MBC value of AgNPs was 15 and 20 µg/ml, respectively. The obtained result agreed with Irrayif et al. (2015), who found that AgNPs at 10 nm size were effective towards *Salmonella*. The bactericidal effect of AgNPs depends on the size and shape of the particles, the smaller particle size increases the specific surface area of nanoparticles, allowing greater number of particles attached to a bacterial cell (Panáček et al. 2006; Pal et al., 2007). Loo et al. (2018) recorded that AgNPs; synthesized using pu-erh tea leaves extract with particle size of 4.06 nm, showed MIC against *S. Typhimurium*, and *S. Enteritidis* at 3.9 and MBC at 7.8, 3.9 µg/mL, respectively.

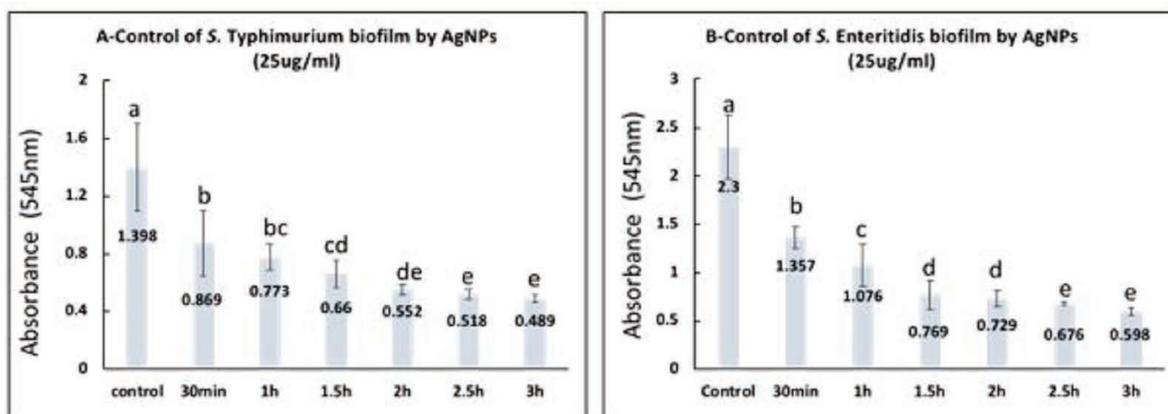


Fig. 5. Control of *S. Typhimurium* (A) and *S. Enteritidis* biofilms (B) by treatment with 25ug/ml AgNPs at different contact times. a,b,c,d,e: means with different letters significantly different.

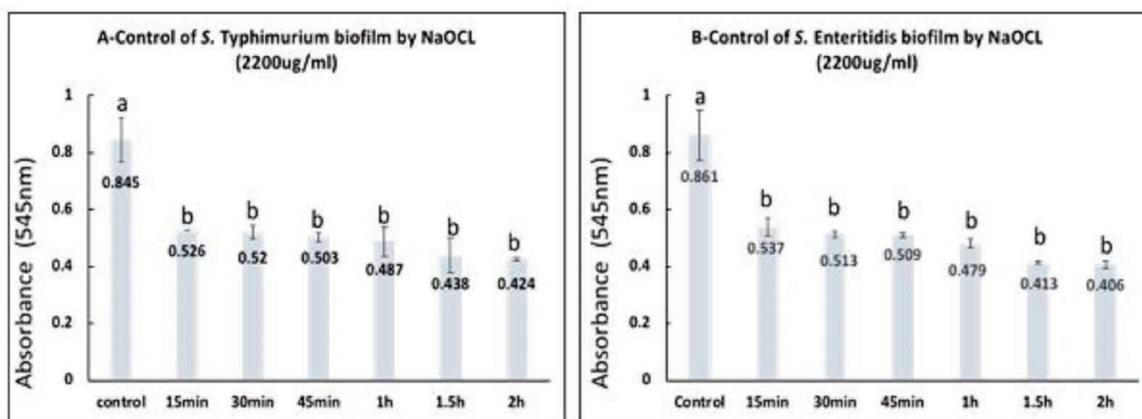


Fig. 6. Control of *S. Typhimurium* (A) and *S. Enteritidis* biofilms (B) by treatment with 2200ug/ml NaOCl at different contact times. a,b: means with different letters are significantly different.

MIC and MBC obtained from this study were higher than that recorded by Samberg *et al.* (2011), where MIC and MBC were 3-4 and 6-12 $\mu\text{g/ml}$ AgNPs (20, 50, 80 μm diameter) against *Salmonella* spp., respectively. Also, no visible growth was observed at concentration 6 $\mu\text{g/ml}$ and MBC was 8 $\mu\text{g/ml}$ AgNPs (75-86 nm diameter) (Gnanadhas *et al.*, 2013). The size of AgNPs in this study is much smaller than that recorded by Samberg *et al.* (2011) and Gnanadhas *et al.* (2013). The antibacterial efficacy of AgNPs depends not only on the different particle sizes but also on surface conditions, furthermore, the surface condition of the particles is directly related to the synthesis method (Samberg *et al.*, 2011). This may explain why result from the present study differs from that of other authors that may be due to the different methods of synthesis. The antimicrobial action of AgNPs may be due to: (1) binding of Ag^+ that prevent the uptake of essential nutrients to bacterial cell, leading to cell death, (2) Ag^+ entry into the cell by competitive binding with essential heavy metals such as Ca^{2+} , Mn^{2+} and Mg^{2+} , or (3) their transport and irreversible accumulation in the cell could occur by complexation with substrates. Finally, Ag^+ could inhibit respiration, or bind and condense DNA once inside (Holt and Bard, 2005). In the end, the antimicrobial activity is likely caused by a synergistic effect between the binding of Ag ions to the cell wall, their uptake and subsequent accumulation in the cell, and their interference with critical biomolecules within the cell.

MIC and MBC of NaOCl were 1600 $\mu\text{g/ml}$ and 2000 $\mu\text{g/ml}$, respectively. These inhibitory levels were lower than that reported by Capita *et al.* (2017), who reported MIC at 6.0 mg/ml. The obtained levels were much higher than that recorded by Corcoran *et al.* (2014) and Espigares *et al.* (2006), who recorded inhibitory levels of 250 and 33-41 $\mu\text{g/ml}$, respectively.

The difference in the inhibitory levels may be attributed to difference in the available chlorine, contact time, method of analysis and different bacterial isolates. Furthermore, the physico-chemical characteristics of NaOCl as pH and surface tension are important for its mechanism of action and antimicrobial activity (Estrela *et al.*, 2002). NaOCl is most widely used for water disinfection and can be used in disinfection of a clinical environment and industrial settings as food, pharmaceuticals, cosmetics and domestic products. NaOCl is an oxidizing biocide and exert its activity through the removal of electrons from sensitive functional groups and the targets can be the cell surface, cell wall or intracellular components giving these agents a very broad spectrum of activity (Estrela *et al.*, 2002). However, NaOCl has some disadvantages as strong odour, decreased efficacy in the presence of organic material, unstable and prone to degradation resulting in formation of toxic by-products (Fukuzaki, 2006).

A significant decrease in biofilm formation by *S. Typhimurium* and *S. Enteritidis* at all AgNPs concentrations was achieved and the anti-biofilms efficiency increased with the increase of concentration, this result agreed with that of Fabrega *et al.* (2009); Kaoud and Yosseif (2013) and Martinez-Gutierrez *et al.* (2013). In the current study, there were no significant differences between biofilm attenuation/prevention effect of various concentration including 10 $\mu\text{g/ml}$ (sub MIC), 15 $\mu\text{g/ml}$ (MIC), 20 $\mu\text{g/ml}$ (MBC) and 25 $\mu\text{g/ml}$, indicating that sub MIC dose of silver does not exhibit cell killing or growth arresting activities but only interfere with biofilm forming ability. The attenuation in biofilm formation could be considered as a potential way to make the bacterial population more susceptible to antimicrobial agents so that they can be removed from the target site (Bjarnsholt *et al.*, 2013). The great effi-

ciency of AgNPs on the prevention of biofilm formation could be explained when the bacterial cells were treated with silver, they attached less to abiotic surfaces and lose their ability to form biofilm (Sharma et al., 2015). Moreover, the antibiofilm activity of AgNPs could be due to inhibition of EPS synthesis that reduces the biofilm formation (McLaughlin-Borlace et al. 1998).

NaOCl was able to promote a significant reduction on the biofilm formation and its efficiency was concentration dependent. This result was in line with the results reported by Russell and McDonnell (2000). During treatment, NaOCl decomposes into sodium hydroxide and hypochlorite, which is a strong oxidizing agent (Tote et al., 2010) that act by oxidizing the cell membrane of bacterium resulting in cell lysis and death (Maris, 1995; Hawkins and Davies, 1999). Although, the chlorinated agents can penetrate into the bacterial biofilm, but these chemicals were not able to completely inactivate all bacteria because the microorganisms in biofilms carry protective mechanisms against the lethal effect of this type of biocidal agents (Stewart et al., 2001).

Studying the effect of AgNPs on biofilms is important because bacteria are often present in biofilm communities. Compared to planktonic cells, biofilm cells have specific biological activities, metabolic pathways and stress responses (Stewart and Franklin, 2008). The EPS of biofilms act as a supporting structure for bacterial adherence and access to nutrients, as well as protects against antimicrobial agents (Høiby et al., 2010). The use of various antimicrobial agents as glutaraldehyde, formaldehyde and peroxygen at 1.0% under field conditions is unsatisfactory to eradicate *Salmonella* biofilms (Marin et al., 2009). The present study revealed that AgNPs were able to remove *Salmonella* spp. biofilm and increasing the contact time between AgNPs and biofilms increased the antibiofilm efficiency and this agreed with Rai et al. (2009). AgNPs are more destructive to biofilms due to better penetration into the EPS matrix (Habash et al., 2014; González et al., 2015). Mature biofilms can tolerate AgNPs by using EPS-mediated trapping, aggregation, and reduced diffusion (Peulen and Wilkinson, 2011; Sheng and Liu, 2011; Joshi et al., 2012). The increased resistance of mature biofilms to antimicrobial agents may be due to several factors. Firstly, bacterial cells in mature biofilms are in the stationary growth phase, therefore, less susceptible to antimicrobial agents (Anderl et al., 2003). Secondly, cells that die in the outer layers of mature biofilms could provide nutrients that enhance the growth of cells in deeper layers (Ito et al., 2009). Thirdly, the high thickness or high amount of EPS in mature biofilms may limit the transport of AgNPs through biofilms. Finally, mature biofilms produce not only more EPS but also other components as, curli which is a protein component used for bacterial adhesion to surfaces (Saldaña et al., 2009). Nevertheless, our results revealed that starch-stabilized AgNPs exhibited not only potent bactericidal activity, but also inhibit the biofilm formation as well as an efficient removal of the mature biofilm formed by *S. Typhimurium* and *S. Enteritidis*.

NaOCl effectively eradicated the mature biofilms formed by *S. Typhimurium* and *S. Enteritidis*. Park et al. (2012) found that 100 mg/L NaOCl resulted in a $\sim 1 \log_{10}$ reduction in the number of cells from *Salmonella* biofilm recovered from steel cells after 5-50 min exposure. Ramesh et al. (2002) mentioned that 0.05% NaOCl was effective and completely eliminated *Salmonella* biofilm as well as produced 7.18 logarithmic reductions in *Salmonella* populations within 2 min., Moretto et al. (2009) found that NaOCl at concentrations of 1300 ppm completely eradicated *Salmonella* biofilms on polystyrene pegs after 1 min of exposure. Vestby et al. (2009) found a 2.4 \log_{10} reduction of 48 h formed *Salmonella enterica* biofilm after treatment by 500 mg/l NaOCl for 5 min. Additionally,

Wong et al. (2010) found that 1 min. exposure *S. Typhimurium* biofilms (3 days old) to NaOCl (1.31 g/l) reduced the bacteria to undetectable levels. Increasing the disinfectant concentration and contact time increased its effectiveness against biofilms (Russell and McDonnell, 2000). Biofilms protects bacteria from detergents and sanitizers, and the bacterial cells are able to develop specific attributes like low multiplication rates or additional defense mechanisms; due to the ability of the extracellular polymeric matrix to neutralize antimicrobial agents, because it consists of organic matter (Rossi and Porto, 2009). Furthermore, mature biofilm provides a more protective factor for bacteria against bactericidal agents. The present study showed that mature biofilms formed by *S. Typhimurium* and *S. Enteritidis* were significantly reduced after treatment with NaOCl.

Conclusion

AgNPs and NaOCl were able to promote significant reduction and control the mature biofilms of *S. Typhimurium* and *S. Enteritidis* isolated from poultry farms. However, the use of AgNPs has the ability to overcome NaOCl disadvantages as odour, instability and decreased efficacy in the presence of organic material. AgNPs is superior to many other antimicrobial agents as they have not been shown to cause bacterial resistance. From these, it can be suggested that AgNPs can be embedded into the matrices or material used for the fabrication of medical devices and poultry equipment to avoid adherence and formation of microbial biofilms. The study also recommends AgNPs compounds to be used as sanitary wipes in food industry and food-related activities. Further research and development are necessary to translate that technology into preventive and therapeutic strategies. NaOCl was also effective in prevention and controlling of biofilms formed by *S. Typhimurium*, *S. Enteritidis* but at higher concentration compared by AgNPs and the results magnified the antibiofilm role of AgNPs compared with NaOCl as 25 mg/l AgNPs was more effective to prevent biofilms than 2200 mg/l NaOCl. So, we can consider AgNPs as a new category of disinfectants that can affect both planktonic and sessile form of bacteria in an excellent manner.

Conflict of interest

Authors declare no conflicts of interest to disclose.

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